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**May 21, 2004**

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FILING DATE UNDER 35 USC 111.**

**APPLICATION NUMBER: 60/542,579**

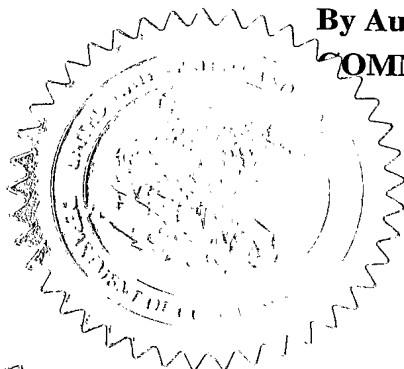
**FILING DATE: February 05, 2004**

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**PROVISIONAL APPLICATION FOR PATENT COVER SHEET**

This is a request for filing a PROVISIONAL APPLICATION FOR PATENT under 37 CFR 1.53(c).

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 Additional inventors are being named on the 1 separately numbered sheets attached hereto**TITLE OF THE INVENTION (500 characters max)**

MESO-SELECTIVE SYNTHESIS OF ANSA-METALLOCENES

Direct all correspondence to:

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**ENCLOSED APPLICATION PARTS (check all that apply)** Specification Number of Pages

25

 CD(s), Number Drawing(s) Number of Sheets Other (specify) Application Data Sheet. See 37 CFR 1.76**METHOD OF PAYMENT OF FILING FEES FOR THIS PROVISIONAL APPLICATION FOR PATENT** Applicant claims small entity status. See 37 CFR 1.27.FILING FEE  
AMOUNT (\$) A check or money order is enclosed to cover the filing fees

\$160.00

 The Commissioner is hereby authorized to charge filing

08-2336

fees or credit any overpayment to Deposit Account Number:

 Payment by credit card. Form PTO-2038 is attached.

The invention was made by an agency of the United States Government or under a contract with an agency of the United States Government.

 No. Yes, the name of the U.S. Government agency and the Government contract number are: \_\_\_\_\_

Respectfully submitted,

SIGNATURE William R. Reid

Date 02/05/2004

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REGISTRATION NO.  
(if appropriate)  
Docket Number.

47,894

TELEPHONE 410-996-1783

LU 6149 (US)

**USE ONLY FOR FILING A PROVISIONAL APPLICATION FOR PATENT**

This collection of information is required by 37 CFR 1.51. The information is used by the public to file (and by the PTO to process) a provisional application. Confidentiality is governed by 35 U.S.C. 122 and 37 CFR 1.14. This collection is estimated to take 8 hours to complete, including gathering, preparing, and submitting the complete provisional application to the PTO. Time will vary depending upon the individual case. Any comments on the amount of time you require to complete this form and/or suggestions for reducing this burden, should be sent to the Chief Information Officer, U.S. Patent and Trademark Office, U.S. Department of Commerce, Washington, D.C. 20231. DO NOT SEND FEES OR COMPLETED FORMS TO THIS ADDRESS. SEND TO: Box Provisional Application, Assistant Commissioner for Patents, Washington, D.C. 20231.

**PROVISIONAL APPLICATION COVER SHEET**  
**Additional Page**

PTO/SB/16 (02-01)

Approved for use through 10/31/2002. OMB 0651-0032  
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Docket Number	LU 6149 (US)
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INVENTOR(S)/APPLICANT(S)		
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Number 2 of 2

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**EXPRESS MAILING CERTIFICATE**

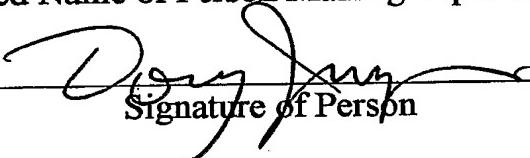
This certifies that the attached **Form PTO/SB/16** (in duplicate), **Specification (25)** for the provisional application of **Valerie Garcia et al.** for **MESO-SELECTIVE SYNTHESIS OF ANSA-METALLOCENES** (our ref: **LU 6149 (US)**) is being mailed by "Express Mail Post Office to Addressee" service in an envelope addressed:

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and deposited on **FEBRUARY 5, 2004** as "Express Mail" in the United States Postal Service, the number of the "Express Mail" mailing label being **ER573871951US**.

Douglas Jumps

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MESO-SELECTIVE SYNTHESIS OF ANSA-METALLOCENES

The present invention relates to a process for the meso-selective preparation of ansa-metallocene complexes of the formula (I), the subsequent reaction of these complexes to form ansa-metallocenes of the formula (IV), the use of transition metal compounds of the formula (III) 5 for preparing metallocenes and also transition metal compounds of the formula (III), ansa-metallocene complexes of the formula (I) and the use of these as constituents of catalyst systems for the polymerization of olefins.

Research and development on the use of organic transition metal compounds, in particular metallocenes, as catalyst components for the polymerization and copolymerization of 10 olefins with the objective of preparing tailored polyolefins has been pursued vigorously in universities and in industry over the past 15 years.

Now, not only the ethylene-based polyolefins prepared by means of metallocene catalyst systems and also, in particular, the propylene-based polyolefins prepared by means of metallocene catalyst systems represent a dynamically growing market segment.

15 To prepare isotactic polypropylenes, ansa-metallocenes in their racemic form are generally used. In the synthesis of the racemic ansa-metallocenes, these are generally obtained together with the undesired meso-metallocenes which are usually separated off without the meso-metallocenes being able to be isolated. It is generally known that ansa-metallocenes in the meso form are responsible for the formation of atactic polypropylenes when used as catalyst 20 constituents.

EP 0 643 078 describes the use of a particular ansa-metallocene in the meso form for preparing very high molecular weight homopolymers and copolymers of ethylene.

While in the case of racemic metallocenes, various racemoselective syntheses have been developed, there has hitherto not yet been an urgent need to develop corresponding meso-selective syntheses.

Organometallics 1997, 16, 5046-49, describes the synthesis of a meso-metallocene or of the corresponding rac-metallocene, in which precisely one of the two possible metallocene forms was formed starting from an isolated diastereomer of a distannylated biscyclopentadienyl ligand system. However, the diastereomerically pure distannylated biscyclopentadienyl ligand system 30 had previously been obtained by separation of the corresponding mixture of diastereomers.

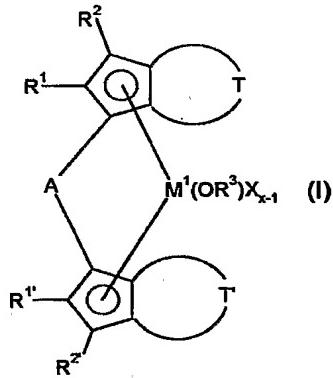
The literature also describes cases in which the rac/meso ratios of the metallocene isomers were dependent on the solvent used in the syntheses starting from zirconium tetrachloride and various dilithiated biscyclopentadienyl ligand systems, but no standard rules were able to be derived.

5 The known methods of preparing particular ansa-metallocenes in the meso form leave something to be desired both in respect of the economics and in respect of the applicability.

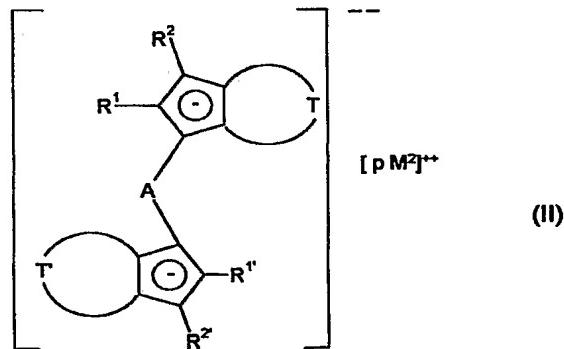
To be able to obtain a better assessment of the potential of ansa-metallocenes in the meso form, there is a need for various meso-metallocenes to be able to be obtained in a simple fashion.

10 It is an object of the present invention to discover a simple, economical and widely applicable process for preparing ansa-metallocenes in the meso form, which offers advantages both in terms of economics and in terms of the applicability.

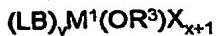
We have found that this object is achieved by a process for the meso-selective preparation of ansa-metallocene complexes of the formula (I),



which comprises reacting a ligand starting compound of the formula (II)



15 with a transition metal compound of the formula (III)



(III)

where

$R^1, R^{1*}$  are identical or different and are each hydrogen or an organic radical having from 1 to 40 carbon atoms,

5     $R^2, R^{2*}$  are identical or different and are each hydrogen or an organic radical having from 1 to 40 carbon atoms,

$R^3$  is a bulky organic radical which has at least 3 carbon atoms, is bound to the oxygen atom via a nonaromatic carbon or silicon atom and may be substituted by halogen atoms or further organic radicals having from 1 to 20 carbon atoms and may also contain

10    heteroatoms selected from the group consisting of Si, N, P, O and S,

$T, T'$  are identical or different and are each a divalent organic group which has from 1 to 40 carbon atoms and together with the cyclopentadienyl ring forms at least one further saturated or unsaturated, substituted or unsubstituted ring system having a ring size of from 5 to 12 atoms, where  $T$  and  $T'$  may contain the heteroatoms Si, Ge, N, P, As, Sb, O, S, Se or Te within the ring system fused to the cyclopentadienyl ring,

15    A is a bridge consisting of a divalent atom or a divalent group,

$M^1$  is an element of group 3, 4, 5 or 6 of the Periodic Table of the Elements or the lanthanides,

the radicals  $x$  are identical or different and are each an organic or inorganic radical which is able 20 to be replaced by a cyclopentadienyl anion,

$x$  is a natural number from 1 to 4,

$M^2$  is an alkali metal, an alkaline earth metal or a magnesium monohalide fragment,

$p$  is 1 in the case of doubly positively charged metal ions or 2 in the case of singly positively charged metal ions or metal ion fragments,

25    LB is an uncharged Lewis base ligand,

and

y is a natural number from 0 to 6.

The radicals  $R^1$  and  $R^{1*}$  are identical or different, preferably identical, and are each hydrogen or an organic radical having from 1 to 40 carbon atoms, for example  $C_1-C_{40}$ -alkyl,  $C_1-$

30     $C_{10}$ -fluoroalkyl,  $C_2-C_{40}$ -alkenyl,  $C_6-C_{40}$ -aryl,  $C_6-C_{10}$ -fluoroaryl, arylalkyl, arylalkenyl or alkylaryl each having from 1 to 10, preferably from 1 to 4, carbon atoms in the alkyl part and

from 6 to 22, preferably from 6 to 10, carbon atoms in the aryl part, or a C<sub>2</sub>-C<sub>40</sub>-heteroaromatic radical which contains at least one heteroatom selected from the group consisting of the elements O, N, S, P and Se, in particular O, N and S, and may be substituted by further radicals R<sup>6</sup>, where R<sup>6</sup> is an organic radical having from 1 to 20 carbon atoms, for example C<sub>1</sub>-C<sub>10</sub>-, preferably C<sub>1</sub>-C<sub>4</sub>-alkyl, C<sub>6</sub>-C<sub>15</sub>-, preferably C<sub>6</sub>-C<sub>10</sub>-aryl, alkylaryl, arylalkyl, fluoroalkyl or fluoroaryl each having from 1 to 10, preferably from 1 to 4, carbon atoms in the alkyl radical and from 6 to 18, preferably from 6 to 10, carbon atoms in the aryl radical, and a plurality of radicals R<sup>6</sup> can be identical or different.

Preference is given to R<sup>1</sup> and R<sup>1'</sup> being identical or different, preferably identical, and each being C<sub>1</sub>-C<sub>10</sub>-alkyl such as methyl, ethyl, n-propyl, isopropyl, n-butyl, i-butyl, s-butyl, t-butyl, n-pentyl, cyclopentyl, n-hexyl, cyclohexyl, n-heptyl or n-octyl, preferably methyl, ethyl or isopropyl, in particular methyl.

The radicals R<sup>2</sup> and R<sup>2'</sup> are identical or different, preferably identical, and are each hydrogen or an organic radical having from 1 to 40 carbon atoms, for example C<sub>1</sub>-C<sub>40</sub>-alkyl, C<sub>1</sub>-C<sub>10</sub>-fluoroalkyl, C<sub>2</sub>-C<sub>40</sub>-alkenyl, C<sub>6</sub>-C<sub>40</sub>-aryl, C<sub>6</sub>-C<sub>10</sub>-fluoroaryl, arylalkyl, arylalkenyl or alkylaryl each having from 1 to 10, preferably from 1 to 4, carbon atoms in the alkyl part and from 6 to 22, preferably from 6 to 10, carbon atoms in the aryl part, or a C<sub>2</sub>-C<sub>40</sub>-heteroaromatic radical which contains at least one heteroatom selected from the group consisting of the elements O, N, S, P and Se, in particular O, N and S, and may be substituted by further radicals R<sup>6</sup>, as defined above, and a plurality of radicals R<sup>6</sup> can be identical or different. Preference is given to R<sup>2</sup> and R<sup>2'</sup> each being hydrogen.

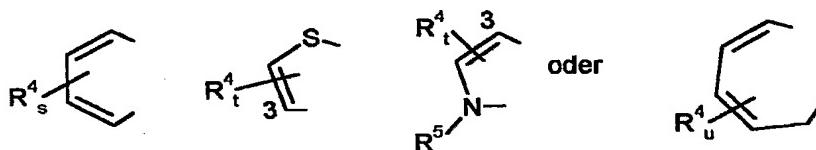
The radical R<sup>3</sup> is a bulky organic radical which has at least 3 carbon atoms, preferably from 4 to 40 carbon atoms, is bound to the oxygen atom via a nonaromatic carbon or silicon atom, preferably a carbon atom, and may be substituted by halogen atoms or further organic radicals having from 1 to 20 carbon atoms and may also contain heteroatoms selected from the group consisting of Si, N, P, O and S, preferably N, O and S. A nonaromatic carbon or silicon atom is an atom of this type which is not located within an aromatic or heteroaromatic ring system, i.e. the oxygen atom of the OR<sup>3</sup> group is not bound directly to an aromatic or heteroaromatic radical.

Examples of such bulky alkyl radicals are isopropyl, cyclohexyl, tert-butyl, 1-adamantyl, 2-adamantyl, triphenylmethyl, diphenylmethyl, (1R)-endo-(+)-1,3,3-trimethyl-2-norbornyl, trimethylsilyl, triphenylsilyl and dimethyl-tert-butylsilyl.

- $R^3$  is preferably an alkali radical which is branched in the  $\alpha$  position and has from 4 to 40, 5 preferably from 7 to 40, carbon atoms and may be substituted by halogen atoms such as fluorine, chlorine, bromine or iodine, in particular fluorine or chlorine, or organic radicals having from 1 to 10 carbon atoms. In this context, an alkyl radical which is branched in the  $\alpha$  position is an alkyl radical whose linking  $\alpha$  atom bears at least two directly bound atoms which are different from hydrogen and not more than one directly bound hydrogen atom.
- 10  $R^3$  is particularly preferably a bicyclic or polycyclic alkyl radical which has from 7 to 30 carbon atoms and may be substituted by one or more  $C_1$ - $C_4$ -alkyl radicals.

- T and  $T'$  are identical or different, preferably identical, and are each a divalent organic group which has from 1 to 40 carbon atoms and together with the cyclopentadienyl ring forms at least one further saturated or unsaturated, substituted or unsubstituted ring system having a ring size of from 5 to 12, in particular from 5 to 7, atoms, where T and  $T'$  may contain the 15 heteroatoms Si, Ge, N, P, As, Sb, O, S, Se or Te, preferably Si, N, O or S, in particular S or N, within the ring system fused to the cyclopentadienyl ring.

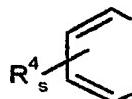
Examples of preferred divalent organic groups T or  $T'$  are



preferably



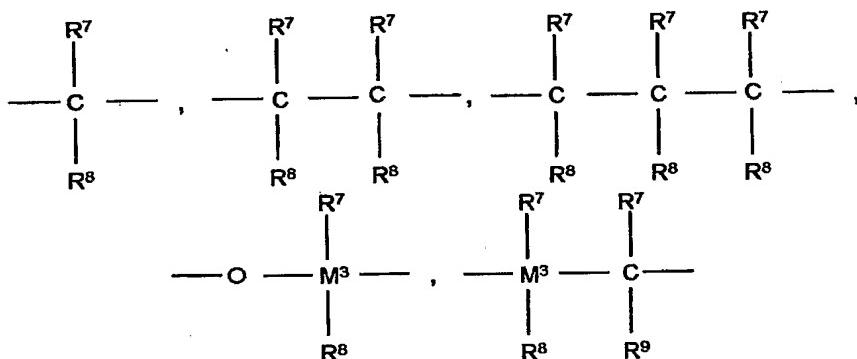
- 20 in particular



where

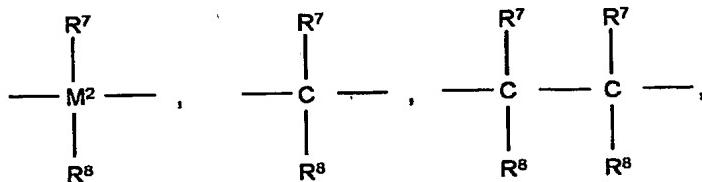
the radicals R<sup>4</sup> are identical or different and are each an organic radical having from 1 to 40, preferably from 1 to 20, carbon atoms, for example cyclic, branched or unbranched C<sub>1</sub>-C<sub>20</sub>-, preferably C<sub>1</sub>-C<sub>8</sub>-alkyl radicals, C<sub>2</sub>-C<sub>20</sub>-, preferably C<sub>2</sub>-C<sub>8</sub>-alkenyl radicals, C<sub>6</sub>-C<sub>22</sub>-, preferably C<sub>6</sub>-C<sub>10</sub>-aryl radicals, alkylaryl or arylalkyl radicals each having from 1 to 10, preferably from 1 to 4, carbon atoms in the alkyl part and from 6 to 22, preferably from 6 to 10, carbon atoms in the aryl part, where the radicals may also be halogenated, or the radicals R<sup>4</sup> are substituted or unsubstituted, saturated or unsaturated, in particular aromatic, heterocyclic radicals which have from 2 to 40, in particular from 4 to 20, carbon atoms and contain at least one heteroatom which is preferably selected from the group consisting of the elements O, N, S and P, in particular O, N and S, or two adjacent radicals R<sup>4</sup> together with the atoms connecting them form a monocyclic or polycyclic, substituted or unsubstituted ring system which has from 1 to 40 carbon atoms and may also contain heteroatoms selected from the group consisting of the elements Si, Ge, N, P, O, S, Se and Te, in particular N or S,

- R<sup>5</sup> is hydrogen or is as defined for R<sup>4</sup>,  
 15 s is a natural number from 0 to 4, in particular from 0 to 3,  
 t is a natural number from 0 to 2, in particular 1 or 2, and  
 u is a natural number from 0 to 6, in particular 1.  
 A is a bridge consisting of a divalent atom or a divalent group. Examples of A are:



- B(R<sup>7</sup>)-, -B(NR<sup>7</sup>R<sup>8</sup>)-, -Al(R<sup>7</sup>)-, -O-, -S-, -S(O)-, -S((O)<sub>2</sub>)-, -N(R<sup>7</sup>)-,  
 20 -C(O)-, -P(R<sup>7</sup>)- or -P(O)(R<sup>7</sup>)-,

in particular



where

M<sup>3</sup> is silicon, germanium or tin, preferably silicon or germanium, particularly preferably silicon, and

R<sup>7</sup>, R<sup>8</sup> and R<sup>9</sup> are identical or different and are each a hydrogen atom, a halogen atom, a trimethylsilyl group, a C<sub>1</sub>-C<sub>10</sub>-, preferably C<sub>1</sub>-C<sub>3</sub>-alkyl group, a C<sub>1</sub>-C<sub>10</sub>-fluoroalkyl group, a C<sub>6</sub>-C<sub>10</sub>-fluoroaryl group, a C<sub>6</sub>-C<sub>10</sub>-aryl group, a C<sub>1</sub>-C<sub>10</sub>-, preferably C<sub>1</sub>-C<sub>3</sub>-alkoxy group, a C<sub>7</sub>-C<sub>15</sub>-alkylaryloxy group, a C<sub>2</sub>-C<sub>10</sub>-, preferably C<sub>2</sub>-C<sub>4</sub>-alkenyl group, a C<sub>7</sub>-C<sub>40</sub>-arylalkyl group, a C<sub>8</sub>-C<sub>40</sub>-arylalkenyl group or a C<sub>7</sub>-C<sub>40</sub>-alkylaryl group or two adjacent radicals together with the atoms connecting them form a saturated or unsaturated ring having from 4 to 15 carbon atoms.

Preferred embodiments of A are the bridges:

dimethylsilanediyl, methylphenylsilanediyl, diphenylsilanediyl, dimethylgermandiyl, ethylidene, 1-ethylethylidene, 1,1-dimethylethylidene, 1,2-dimethylethylidene, 1,1,2,2-tetramethylethylidene, dimethylmethyliidene, phenylmethylmethyliidene or diphenylmethyliidene, in particular dimethylsilanediyl, diphenylsilanediyl and ethylidene.

A is particularly preferably a substituted silylene group or a substituted or unsubstituted ethylene group, preferably a substituted silylene group such as dimethylsilanediyl, methylphenylsilanediyl, methyl-tert-butylsilanediyl or diphenylsilanediyl, in particular dimethylsilanediyl.

M<sup>1</sup> is an element of group 3, 4, 5 or 6 of the Periodic Table of the Elements or the lanthanides, for example titanium, zirconium, hafnium, vanadium, niobium, tantalum, chromium, molybdenum or tungsten, preferably titanium, zirconium or hafnium, particularly preferably zirconium or hafnium and very particularly preferably zirconium.

The radicals X are identical or different and are each an organic or inorganic radical which is able to be replaced by a cyclopentadienyl anion. Examples of X are halogen such as chlorine, bromine, iodine, in particular chlorine, organosulfonate groups such as trifluoromethanesulfonate (triflate) or mesylate. X is preferably halogen, in particular chlorine.

x is a natural number from 1 to 4 and usually corresponds to the oxidation number of M<sup>1</sup> minus 2. In the case of elements of group 4 of the Periodic Table of the Elements, x is preferably 2.

M<sup>2</sup> is an alkali metal such as Li, Na or K, an alkaline earth metal such as Mg or Ca, in particular Mg, or a magnesium monohalide fragment such as MgCl, MgBr or MgI. M<sup>2</sup> is preferably Li, Na, K, MgCl, MgBr, MgI or Mg, particularly preferably Li, K or Mg, in particular Li.

p is 1 for doubly positively charged metal ions or 2 for singly positively charged metal ions or metal ion fragments.

LB is an uncharged Lewis base ligand, preferably a linear, cyclic or branched oxygen-, sulfur, nitrogen- or phosphorus-containing, in particular oxygen- or nitrogen-containing, hydrocarbon such as an ether, polyether, thioether, amine, polyamine or phosphine. LB is preferably a cyclic or acyclic ether or diether, for example diethyl ether, dibutyl ether, tert-butyl methyl ether, anisole, dimethoxyethane (DME), tetrahydrofuran (THF) or dioxane. Particular preference is given to THF or DME.

y is a natural number from 0 to 6. In the case of elements of group 4 of the Periodic Table of the Elements, y is preferably 1 or 2.

Furthermore, the substituents according to the present invention are, unless restricted further, defined as follows:

The term "organic radical having from 1 to 40 carbon atoms" as used in the present context refers, for example, to C<sub>1</sub>-C<sub>40</sub>-alkyl radicals, C<sub>1</sub>-C<sub>10</sub>-fluoroalkyl radicals, C<sub>1</sub>-C<sub>12</sub>-alkoxy radicals, saturated C<sub>3</sub>-C<sub>20</sub>-heterocyclic radicals, C<sub>6</sub>-C<sub>40</sub>-aryl radicals, C<sub>2</sub>-C<sub>40</sub>-heteroaromatic radicals, C<sub>6</sub>-C<sub>10</sub>-fluoroaryl radicals, C<sub>6</sub>-C<sub>10</sub>-aryloxy radicals, C<sub>3</sub>-C<sub>18</sub>-trialkylsilyl radicals, C<sub>2</sub>-C<sub>20</sub>-alkenyl radicals, C<sub>2</sub>-C<sub>20</sub>-alkynyl radicals, C<sub>7</sub>-C<sub>40</sub>-arylalkyl radicals or C<sub>8</sub>-C<sub>40</sub>-arylalkenyl radicals. An organic radical is in each case derived from an organic compound. Thus, the organic compound methanol can in principle give rise to three different organic radicals having one carbon atom, namely methyl (H<sub>3</sub>C-), methoxy (H<sub>3</sub>C-O-) and hydroxymethyl (HOC(H<sub>2</sub>)-).

The term "alkyl" as used in the present context encompasses linear or singly or multiply branched saturated hydrocarbons which may also be cyclic. Preference is given to a C<sub>1</sub>-C<sub>18</sub>-alkyl group such as methyl, ethyl, n-propyl, n-butyl, n-pentyl, n-hexyl, n-heptyl, n-octyl, n-nonyl, n-decyl, cyclopentyl, cyclohexyl, isopropyl, isobutyl, isopentyl, isohexyl, sec-butyl or tert-butyl.

The term "alkenyl" as used in the present context encompasses linear or singly or multiply branched hydrocarbons having at least one, if desired more than one, C-C double bonds which may be cumulated or alternating.

The term "saturated heterocyclic radical" as used in the present context refers, for example, to monocyclic or polycyclic, substituted or unsubstituted hydrocarbon radicals in which one or more carbon atoms, CH groups and/or CH<sub>2</sub> groups have been replaced by heteroatoms which are preferably selected from the group consisting of O, S, N and P. Preferred examples of substituted or unsubstituted saturated heterocyclic radicals are pyrrolidinyl, imidazolidinyl, pyrazolidinyl, piperidyl, piperazinyl, morpholinyl, tetrahydrofuranlyl, tetrahydropyranyl, tetrahydrothienyl and the like, and also methyl-, ethyl-, propyl-, isopropyl- and tert-butyl-substituted derivatives thereof.

The term "aryl" as used in the present context refers, for example, to aromatic and fused or unfused polyaromatic hydrocarbon substituents which may be monosubstituted or polysubstituted by linear or branched C<sub>1</sub>-C<sub>18</sub>-alkyl, C<sub>1</sub>-C<sub>18</sub>-alkoxy, C<sub>2</sub>-C<sub>10</sub>-alkenyl or halogen, in particular fluorine. Preferred examples of substituted and unsubstituted aryl radicals are, in particular, phenyl, pentafluorophenyl, 4-methylphenyl, 4-ethylphenyl, 4-n-propylphenyl, 4-isopropylphenyl, 4-*tert*-butylphenyl, 4-methoxyphenyl, 1-naphthyl, 9-anthryl, 9-phenanthryl, 3,5-dimethylphenyl, 3,5-di-*tert*-butylphenyl and 4-trifluoromethylphenyl.

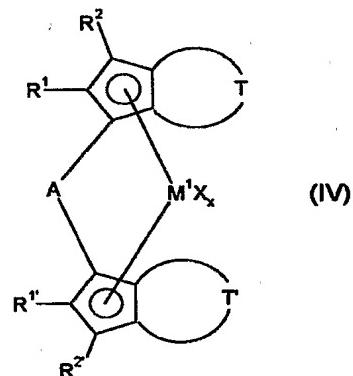
The term "heteroaromatic radical" as used in the present context refers, for example, to aromatic hydrocarbon radicals in which one or more carbon atoms have been replaced by nitrogen, phosphorus, oxygen or sulfur atoms or combinations thereof. They may, like the aryl radicals, be monosubstituted or polysubstituted by linear or branched C<sub>1</sub>-C<sub>18</sub>-alkyl, C<sub>2</sub>-C<sub>10</sub>-alkenyl or halogen, in particular fluorine. Preferred examples are furyl, thienyl, pyrrolyl, pyridyl, pyrazolyl, imidazolyl, oxazolyl, thiazolyl, pyrimidinyl, pyrazinyl and the like, and also methyl-, ethyl-, propyl-, isopropyl- and *tert*-butyl-substituted derivatives thereof.

The term "arylalkyl" as used in the present context refers, for example, to aryl-containing substituents whose aryl radical is connected via an alkyl chain to the remainder of the molecule. Preferred examples are benzyl, substituted benzyl, phenethyl, substituted phenethyl and the like.

The terms fluoroalkyl and fluoroaryl indicate that at least one hydrogen atom, preferably two or more and at most all hydrogen atoms, of the respective substituent have been replaced by

fluorine atoms. Examples of fluorine-containing substituents which are preferred according to the present invention are trifluoromethyl, 2,2,2-trifluoroethyl, pentafluorophenyl, 4-trifluoromethylphenyl, 4-perfluoro-tert-butylphenyl and the like.

In a preferred embodiment of the process of the present invention, the metallocene complex of the formula (I) is converted into an ansa-metallocene complex of the formula (IV)



where the variables and indices have the same meanings as in the formula (I), by reaction with suitable elimination reagents in a subsequent reaction step.

The present invention therefore also provides for the use of a metallocene complex of the formula (I) as intermediate for preparing ansa-metallocene complexes of the formula (IV).

Elimination reagents are known in principle. Examples of preferred elimination reagents are hydrogen halides, for example HCl, and aliphatic or aromatic carboxylic halides, for example acetyl chloride, acetyl bromide, phenylacetyl chloride, tert-butylacetyl chloride, and also organoaluminum halides, for example ethylaluminum dichloride, methylaluminum dichloride or dimethylaluminum chloride, and halogen-containing main group compounds such as SiCl<sub>4</sub>,

SOCl<sub>2</sub>, PCl<sub>5</sub> or AlCl<sub>3</sub>.

Particularly preferred elimination reagents are HCl, acetyl chloride, ethylaluminum dichloride and methylaluminum dichloride.

The elimination reaction is usually carried out in a temperature range from 0°C to 110°C. To complete the reaction, it is usual to use at least stoichiometric amounts of the elimination 20 reagent. Excess elimination reagent generally does not interfere as long as it can be separated off from the target product without problems in the work-up.

Particular preference is given to a process for the meso-selective preparation of ansa-metallocene complexes of the formula (I) which optionally further comprises the subsequent reaction of these complexes to form ansa-metallocenes of the formula (IV)  
in which

- 5     $R^1, R^{1'}$  are identical or different and are each  $C_1-C_{10}$ -alkyl such as methyl, ethyl, isopropyl, tert-butyl, cyclohexyl or octyl, in particular methyl,  
 $R^2, R^{2'}$  are each hydrogen,  
T,  $T'$  are identical or different and are each an unsubstituted 1,3-butadiene-1,4-diyl group or a  
1,3-butadiene-1,4-diyl group substituted by from 1 to 4 radicals  $R^4$ , where  $R^4$  can be  
10    identical or different and are organic radicals having from 1 to 40 carbon atoms,  
A    is ethylene, substituted ethylene or substituted silylene, in particular substituted silylene  
such as dimethylsilanediyl, methylphenylsilanediyl, methyl-tert-butylsilanediyl or  
diphenylsilanediyl, in particular dimethylsilanediyl,  
and the variables  $R^3, M^1, X, M^2$  and LB and also the indices x, p and y are as defined for the  
15    formula I.

Very particular preference is given to a process for the meso-selective preparation of  
ansa-metallocene complexes of the formula (I) or the formula (IV), as described above,  
in which

- 20     $R^3$  is an alkyl radical which is branched in the  $\alpha$  position and has from 4 to 40, preferably  
from 7 to 40, carbon atoms and may be substituted by halogen atoms such as fluorine,  
chlorine, bromine or iodine, in particular fluorine or chlorine, or by organic radicals  
having from 1 to 10 carbon atoms, with  $R^3$  particularly preferably being a bicyclic or  
polycyclic alkyl radical which has from 7 to 30 carbon atoms and may be substituted by  
one or more  $C_1-C_4$ -alkyl radicals,  
25     $M^1$  is Ti, Zr or Hf, preferably Zr or Hf, in particular Zr,  
X is halogen, in particular chlorine,  
x is 2,  
LB is a cyclic or acyclic ether or diether, in particular THF or DME,  
and  
30    y is 1 or 2.

In the process of the present invention, the salt-like ligand starting compounds of the formula (II) can either be used in isolated form or be prepared in-situ immediately before the reaction with the transition metal compound of the formula (III).

To synthesize the salt-like ligand starting compounds of the formula (II), the corresponding uncharged bridged biscyclopentadienyl compound is usually doubly deprotonated by means of a strong base. Strong bases which can be used are, for example, organometallic compounds or metal hydrides, preferably compounds containing an alkali metal or an alkaline earth metal. Preferred bases are organolithium or organomagnesium compounds such as methylolithium, n-butyllithium, sec-butyllithium, n-butyl-n-octylmagnesium or dibutylmagnesium.

The uncharged bridged biscyclopentadienyl compound to be deprotonated can also be used in isolated form or without isolation, prepared directly by the coupling reaction of two cyclopentadienyl anions with an appropriate bridging reagent, for example a diorganodichlorosilane such as dimethyldichlorosilane. A further possible way of preparing the uncharged biscyclopentadienyl compounds is stepwise construction. Here, for example, a cyclopentadienyl anion is firstly reacted with an appropriate bridging reagent, for example a diorganodichlorosilane such as dimethyldichlorosilane, to form a monochloromonocyclopentadienyldiorganosilane compound and the chlorine in this is subsequently replaced by a further cyclopentadienyl radical, which may be different from the first cyclopentadienyl radical, to obtain the desired uncharged bridged biscyclopentadienyl compound.

The synthesis of the cyclopentadienyl anions can in principle be carried out under the same conditions as the deprotonation of the uncharged bridged biscyclopentadienyl compound.

The double deprotonation of the uncharged bridged biscyclopentadienyl compound to form the ligand starting compound of the formula (II) is usually carried out at from -78°C to 110°C, preferably from 0°C to 80°C and particularly preferably from 20°C to 70°C.

Suitable inert solvents in which the deprotonation of the cyclopentadienyl derivatives by means of strong bases can be carried out are aliphatic or aromatic hydrocarbons such as benzene, toluene, xylene, mesitylene, ethylbenzene, cumene, decalin, tetralin, pentane, hexane, cyclohexane, heptane or ethers such as diethyl ether, di-n-butyl ether, tert-butyl methyl ether (MTBE), tetrahydrofuran (THF), 1,2-dimethoxyethane (DME), anisole, triglyme, dioxane and

any mixtures of these substances. Preference is given to solvents or solvent mixtures in which the subsequent process of the present invention for preparing the metallocene complexes of the formula (I) can likewise be carried out directly.

The synthesis of the transition metal compounds of the formula (III) is in principle known from the literature. A possible way of preparing them is, for example, reaction of a transition metal compound  $M^1X_{x+2}$  or  $(LB)_yM^1X_{x+2}$  with a metal alkoxide  $M^2(OR^3)$  in an inert solvent, where  $M^2$  and  $M^1$  and the other variables are as defined for the formula (I).

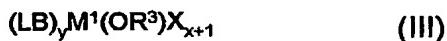
In the process of the present invention, the reaction of the ligand starting compound of the formula (II) with the transition metal compound of the formula (III) can be carried out in an inert solvent or solvent mixture at from  $-78^\circ C$  to  $150^\circ C$ , in particular from  $0^\circ C$  to  $110^\circ C$ . The inert solvents or solvent mixtures which can be used are preferably the same ones in which the synthesis of the ligand starting compound of the formula (II) has been carried out. The reaction times are usually at least 10 minutes, generally from 1 to 8 hours.

The present invention therefore also provides for the use of a transition metal compound of the formula (III)



for the preparation of ansa-metallocene complexes, in particular for the meso-selective preparation of ansa-metallocene complexes of the formula (I) or for the preparation of ansa-metallocene complexes of the formula (IV) by the process of the present invention, where the variables and indices are as described above.

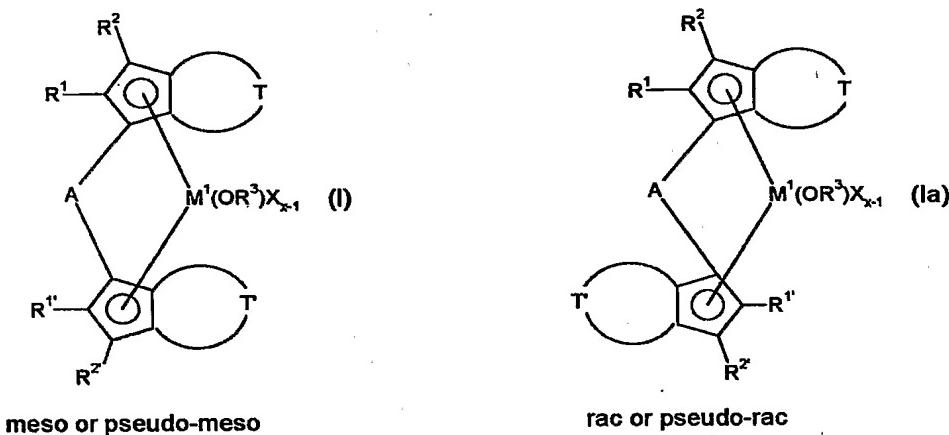
The present invention further provides transition metal compounds of the formula (III)



where the variables and indices are as described above.

Particular preference is given to compounds of the formula (III) in which  $R^3$  is an alkyl radical which is branched in the  $\alpha$  position and has from 4 to 40, in particular from 7 to 40, carbon atoms and may be substituted by halogen atoms or organic radicals having from 1 to 10 carbon atoms, in particular compounds in which  $R^3$  is a bicyclic or polycyclic alkyl radical having from 7 to 30 carbon atoms and optionally bearing one or more  $C_1-C_4$ -alkyl radicals as substituents and in which  $M^1$  is Ti, Zr or Hf, in particular Zr or Hf, and in which X is halogen, in particular chlorine, x is 2, LB is a cyclic or acyclic ether or diether and y is 1 or 2.

In the process of the present invention, it is possible for not only the desired meso compounds of the formula (I) but also the corresponding rac compounds to be formed, where the terms meso and rac refer to the spatial arrangement of the two cyclopentadienyl ring systems relative to one another. For example, in cases in which the two cyclopentadienyl radicals on the bridge are not identical, there is no meso form having  $C_s$  symmetry or rac form having  $C_2$  symmetry, but instead there are only diastereomeric compounds having  $C_1$  symmetry. These various diastereomeric metallocene compounds which differ from one another in the spatial arrangement of the different substituents behave, when used as catalyst components in the polymerization of propylene, like the  $C_s$ -symmetric meso isomer (atactic polypropylene) or like the  $C_2$ -symmetric rac isomer (isotactic polypropylene) simply on the basis of the spatial arrangement of the two cyclopentadienyl ligands relative to one another and can thus be referred to as either a pseudo-rac form or a pseudo-meso form.



In the following, meso and pseudo-meso form and rac and pseudo-rac form are distinguished simply as rac and meso forms.

Furthermore, the meso selectivity = (proportion of meso – proportion of rac)/(proportion of rac + proportion of meso) in the process of the present invention is greater than zero, preferably greater than 0.5.

The salts of the formulae  $M^2X$  or  $M^2X_2$ , for example lithium chloride or magnesium chloride, which are obtained as further reaction product in the process of the present invention for preparing meso ansa-metallocenes of the formula (I) can be separated off from the metallocene by known methods. For example, a salt such as lithium chloride can be precipitated by means of a suitable solvent in which the metallocene is, however, soluble, so that the solid

lithium chloride is separated off from the dissolved metallocene by means of a filtration step. The metallocene can also be separated off from the salt by extraction with such a solvent. If filtration steps are employed, use can also be made of filter aids such as kieselguhr. Organic solvents which are suitable for such a filtration or extraction step are, in particular, organic  
5 aprotic, oxygen-free solvents such as toluene, ethylbenzene, xylenes and methylene chloride. If appropriate, the solvent constituents in which the salt is at least partially soluble are largely removed before the above-described removal of the salt. For example, lithium chloride is appreciably soluble in tetrahydrofuran. For this reason, an alternative is to remove the salts of the formulae  $M^2X$  and  $M^2X_2$  with the aid of a solvent or solvent mixture in which they are  
10 readily soluble while the metallocene complex is sparingly soluble therein.

The ansa-metallocene complexes of the formula (I) prepared by the process of the present invention are used, together with suitable cocatalysts and, if appropriate, suitable support materials, as constituents of a catalyst system for the polymerization of olefins.

The present invention further provides ansa-metallocene complexes of the formula (I) as are obtainable by the process of the present invention and also the use of an ansa-metallocene complex of the formula (I) prepared by a process as claimed in claim 1 as constituent of a catalyst system for the polymerization of olefins.  
15

Preference is given to ansa-metallocene complexes of the formula (I) as described above in which  $R^3$  is a bicyclic or polycyclic alkyl radical which has from 7 to 30 carbon atoms and  
20 may bear one or more  $C_1$ - $C_4$ -alkyl radicals as substituents,  $M^1$  is Ti, Zr or Hf, in particular Zr or Hf, X is halogen, in particular chlorine, and x is 2.

Preference is given to metallocene mixtures which are obtainable directly by the process of the present invention and comprise more than 50 mol% of metallocenes of the formula (I) and less than 50 mol% of metallocenes of the formula (Ia), based on the total amount of metallocene  
25 compounds. Particular preference is given to mixtures which comprise more than 75 mol% of metallocenes of the formula (I) and less than 25 mol% of metallocenes of the formula (Ia).

The invention is illustrated by the following nonrestrictive examples:

### Examples

#### **General procedures:**

Synthesis and handling of the organometallic compounds was carried out in the absence  
30 of air and moisture under argon (glove box and Schlenk technique). All solvents used were

purged with argon and dried over molecular sieves before use. NMR spectra of organic and organometallic compounds were recorded at room temperature on a Varian Unity-300 NMR spectrometer. The chemical shifts are reported relative to SiMe<sub>4</sub>.

1. **Synthesis of meso-dimethylsilanediylbis(2-methylindenyl)zirconium dichloride via  
5 intermediate isolation of meso-dimethylsilanediylbis(2-methylindenyl)zirconium monochloride 1-adamantoxide**

**1a Synthesis of ZrCl<sub>4</sub>(THF)<sub>2</sub>**

Under protective gas, 15.2 g of zirconium tetrachloride were suspended in 80 g of dry toluene. The suspension was cooled in an ice bath and 12 g of tetrahydrofuran (THF) were added slowly. The colorless suspension was subsequently stirred at room temperature for a further one hour.

**1b Synthesis of lithium 1-adamantoxide**

A solution of 9.6 g of 1-adamantanone in 80 g of toluene and 12 g of THF was cooled in an ice bath and 21.2 g of a solution of n-butyllithium (20% by weight in toluene) were added dropwise over a period of half an hour. The reaction mixture was subsequently stirred at room temperature for a further hour.

**1c Synthesis of (THF)<sub>2</sub>Cl<sub>3</sub>Zr(1-adamantoxide)**

The solution of lithium 1-adamantoxide prepared in example 1b was added dropwise at room temperature to the suspension of the zirconium tetrachloride-THF complex prepared in example 1a over a period of 20 minutes. The reaction mixture was stirred at room temperature for a further 2 hours.

**1d Synthesis of Li<sub>2</sub>[dimethylsilanediylbis(2-methylindenyl)]**

42 g of a solution of n-butyllithium (20% by weight in toluene) were added at 0-4°C to a solution of 20 g of dimethylbis(2-methylindenyl)silane (63.30 mmol) in 132 g of toluene and 25 12 g of THF over a period of 30 minutes. A yellowish beige suspension was formed and this was stirred at room temperature for a further 1.5 hours.

**1e Synthesis and isolation of meso-dimethylsilanediylbis(2-methylindenyl)zirconium monochloride 1-adamantoxide (1e)**

The suspension of Li<sub>2</sub>[dimethylsilanediylbis(2-methylindenyl)] prepared in example 1d was added at room temperature to the suspension of (THF)<sub>2</sub>Cl<sub>3</sub>Zr(1-adamantoxide), resulting in immediate formation of a yellow suspension. The suspension was stirred at room temperature

for a further 2.5 hours and subsequently filtered through a G4 frit. The filtercake was extracted three times with 200 g each time of toluene. The combined filtrates were concentrated under reduced pressure to 12% of the initial mass and allowed to stand at -20°C for 16 hours. The yellow precipitate formed was filtered off, washed with 15 ml of toluene and dried under reduced pressure. This gave 6.16 g of (1e) as a yellow powder. The orange filtercake which remained after the toluene extraction was extracted with 200 g of dichloromethane. Removal of the solvent under reduced pressure once again gave a yellow powder which was washed with 30 g of heptane and subsequently dried in an oil pump vacuum. A further 11.36 g of complex (1e) were obtained. Overall, the complex (1e) was obtained in a yield of 46.7% (17.52 g) in the form of a yellow powder.

<sup>1</sup>H-NMR (CD<sub>2</sub>Cl<sub>2</sub>): 7.72 (dd, J = 8.7 Hz and 1.0 Hz, 2H, aromatic), 7.22-7.19 (m, 2H, aromatic), 7.00-6.96 (m, 2H, aromatic), 6.65-6.61 (m, 2H, aromatic), 6.25 (s, 1H, Cp), 2.52 (s, 6H, 2 x CH<sub>3</sub>-Cp), 2.15-2.12 (m, 3H, adamantyl), 1.80-1.79 (m, 6H, adamantyl), 1.50-1.48 (m, 6H, adamantyl), 1.37 (s, 3H, CH<sub>3</sub>Si), 1.20 (s, 3H, CH<sub>3</sub>Si).

**1. Synthesis of meso-dimethylsilanediylbis(2-methylindenyl)zirconium dichloride (1)**

2 g of acetyl chloride were added to a suspension of 10 g of the complex (1e) prepared in example 1e in 100 ml of n-heptane. The reaction mixture was stirred at 40°C for 5 hours. The orange suspension was filtered through a G3 frit, the filtercake was washed with 30 g of n-heptane and 20 g of toluene and dried under reduced pressure. This gave 5.12 g of the compound (1) as an orange powder (64% yield based on (1e)).

<sup>1</sup>H-NMR (CD<sub>2</sub>Cl<sub>2</sub>): 7.67 (dd, J = 8.8Hz and 0.9Hz, 2H, aromatic), 7.38-7.36 (m, 2H, aromatic), 7.11-7.07 (m, 2H, aromatic), 6.77-6.73 (m, 2H, aromatic), 6.65 (s, 1H, Cp), 2.44 (s, 6H, 2 x CH<sub>3</sub>-Cp), 1.43 (s, 3H, CH<sub>3</sub>Si), 1.23 (s, 3H, CH<sub>3</sub>Si).

**2. Single-vessel synthesis of meso-enriched dimethylsilanediylbis(2-methylindenyl)zirconium dichloride (1) without intermediate isolation of dimethylsilanediylbis(2-methylindenyl)zirconium monochloro 1-adamantoxide**

A suspension of (THF)<sub>2</sub>Cl<sub>3</sub>Zr(1-adamantoxide) which had been prepared in the same amount and in the same way as in example 1c was added to a suspension of Li<sub>2</sub>[dimethylsilanediylbis(2-methylindenyl)] which had been prepared as in example 1d and the mixture was stirred at room temperature for 2.5 hours. The yellow suspension was filtered through a G4 glass frit filter and the filtercake was extracted three times with 200 g each time of

toluene. The filtrate was evaporated to 50% of its original weight. 4.98 g of acetyl chloride were added to the solution and the reaction mixture was stirred at 40°C for 3 hours. The reaction solution was concentrated to 10% of its initial mass. The precipitate formed was filtered off with the aid of a G3 glass frit filter, washed with 10 g of toluene and dried under reduced pressure.

- 5 This gave 4 g of the compound (1) in the form of an orange powder (14% yield), with the rac/meso ratio as determined by <sup>1</sup>H-NMR being 1:4.

10 3. **Single-vessel synthesis of meso-enriched dimethylsilanediylbis(2-methylindenyl)zirconium dichloride (1) without intermediate isolation of meso-enriched dimethylsilanediylbis(2-methylindenyl)zirconium monochloride endo-(-)-fenchoxide**

The experiment was carried out in a manner analogous to example 2. ZrCl<sub>4</sub>(THF)<sub>2</sub> was prepared in the same amount as in example 1a and combined with a solution of lithium endo-(-)-fenchoxide which had been prepared from 9.76 g of endo-(-)-fenchol and 21.2 g of a solution of n-butyllithium (20% by weight in toluene) by a method analogous to example 1b, giving a suspension of (THF)<sub>2</sub>Cl<sub>3</sub>Zr(fenchoxide). This suspension was added to a suspension of Li<sub>2</sub>[dimethylsilanediylbis(2-methylindenyl)] which had been prepared as described in example 1d from 20 g of dimethylbis(2-methylindenyl)silane (63.30 mmol). This resulted in immediate formation of a yellow suspension which was stirred at room temperature for 1.5 hours. The yellow suspension was filtered through a G4 glass frit filter and the filtercake was extracted 3 times with 200 g each time of toluene. The filtrate was evaporated to 12% of its initial mass. Some yellow crystals were able to be isolated, and these were examined by NMR spectroscopy and found to be dimethylsilanediylbis(2-methylindenyl)zirconium monochloride endo-(-)-fenchoxide.

15 <sup>1</sup>H-NMR (CD<sub>2</sub>Cl<sub>2</sub>: 7.66-7.63 (m, 2H, aromatic), 7.27-7.15 (2dd, 2H, aromatic), 7.05-6.98 (m, 2H, aromatic), 6.69-6.65 (m, 2H, aromatic), 6.58 and 6.36 (2s, 2H, 2 x Cp), 3.91 (s, 1H, fenchyl), 2.53 and 2.49 (2s, 6H, 2 x CH<sub>3</sub>-Cp), 1.36, 1.20 and 1.18 (3s, 9H, CH<sub>3</sub>-fenchyl), 1.03 (s, 3H, CH<sub>3</sub>Si), 0.92 (s, 3H, CH<sub>3</sub>Si), 1.69-1.30, 1.10-1.04, 0.98 and 0.85-0.77 (m, H, fenchyl).

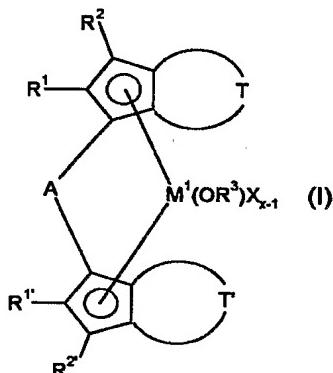
20 4.97 g of acetyl chloride were added to the partly evaporated solution and the reaction mixture was stirred at 40°C for one hour. The precipitate was filtered off with the aid of a G3 glass frit filter, washed with 40 g of n-heptane and dried under reduced pressure. This gave

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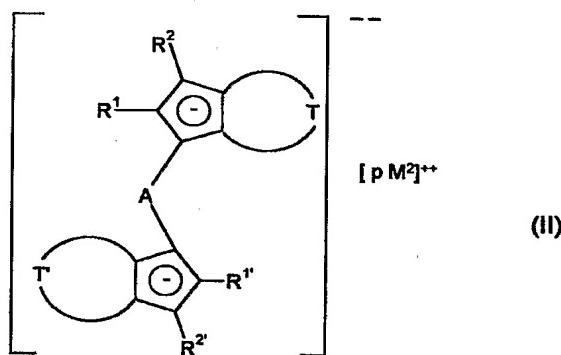
4.51 g of the compound (1) in the form of an orange powder (14% yield), with the rac/meso ratio as determined by  $^1\text{H-NMR}$  being 1:6.

We claim:

1. A process for the meso-selective preparation of ansa-metallocene complexes of the formula (I),



which comprises reacting a ligand starting compound of the formula (II)



5

with a transition metal compound of the formula (III)



where

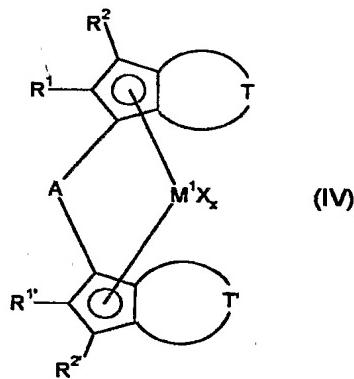
$R^1, R^{1'}$  are identical or different and are each hydrogen or an organic radical having from 1 to 40 carbon atoms,

10  $R^2, R^{2'}$  are identical or different and are each hydrogen or an organic radical having from 1 to 40 carbon atoms,

$R^3$  is a bulky organic radical which has at least 3 carbon atoms, is bound to the oxygen atom via a nonaromatic carbon or silicon atom and may be substituted by halogen atoms or further organic radicals having from 1 to 20 carbon atoms and

may also contain heteroatoms selected from the group consisting of Si, N, P, O and S,

- 5      T, T' are identical or different and are each a divalent organic group which has from 1 to 40 carbon atoms and together with the cyclopentadienyl ring forms at least one further saturated or unsaturated, substituted or unsubstituted ring system having a ring size of from 5 to 12 atoms, where T and T' may contain the heteroatoms Si, Ge, N, P, As, Sb, O, S, Se or Te within the ring system fused to the cyclopentadienyl ring,
- 10     A    is a bridge consisting of a divalent atom or a divalent group,
- M<sup>1</sup>    is an element of group 3, 4, 5 or 6 of the Periodic Table of the Elements or the lanthanides,
- the radicals X are identical or different and are each an organic or inorganic radical which is able to be replaced by a cyclopentadienyl anion,
- 15     x    is a natural number from 1 to 4,
- M<sup>2</sup>    is an alkali metal, an alkaline earth metal or a magnesium monohalide fragment,
- p    is 1 in the case of doubly positively charged metal ions or 2 in the case of singly positively charged metal ions or metal ion fragments,
- LB    is an uncharged Lewis base ligand,
- and
- 20     y    is a natural number from 0 to 6.
2.    A process as claimed in claim 1, wherein the metallocene complex of the formula (I) is converted into an ansa-metallocene complex of the formula (IV)



where

the variables and indices have the same meanings as in the formula (I), by reaction with suitable elimination reagents in a subsequent reaction step.

3. A process as claimed in claim 1 or 2, wherein

$R^1, R^{1'}$  are identical or different and are each  $C_1-C_{10}$ -alkyl,

5  $R^2, R^{2'}$  are each hydrogen,

$T, T'$  are identical or different and are each an unsubstituted 1,3-butadiene-1,4-diyil group or a 1,3-butadiene-1,4-diyil group substituted by from 1 to 4 radicals  $R^4$ , where  $R^4$  can be identical or different and are organic radicals having from 1 to 40 carbon atoms,

10 A is ethylene, substituted ethylene or substituted silylene,

and the variables  $R^3, M^1, X, M^2$  and LB and also the indices x, p and y are as defined in claim 1.

4. A process as claimed in any of claims 1 to 3, wherein

$R^3$  is an alkyl radical which is branched in the  $\square$  position and has from 4 to 40 carbon atoms and may be substituted by halogen atoms or organic radicals having from 1 to 10 carbon atoms,

$M^1$  is Ti, Zr or Hf,

$X$  is halogen,

x is 2,

20 LB is a cyclic or acyclic ether or diether,

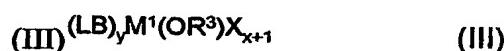
and

y is 1 or 2.

5. A process as claimed in any of claims 1 to 4, wherein

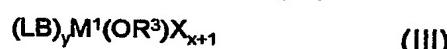
$M^2$  is Li, Na, K, MgCl, MgBr, MgI or Mg.

- 25 6. The use of a transition metal compound of the formula



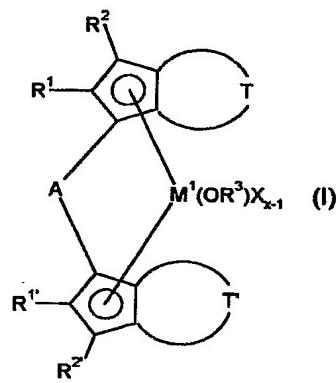
for preparing ansa-metallocene complexes.

7. A transition metal compound of the formula (III)



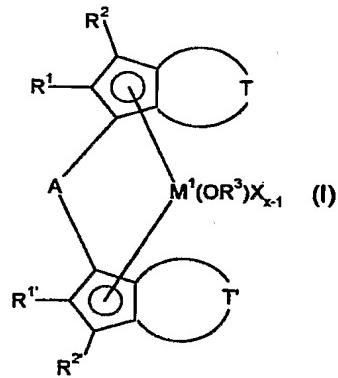
30 where the variables and indices are as defined in claim 1 or claim 4.

8. The use of a metallocene complex of the formula (I) as set forth in claim 1



as intermediate for preparing ansa-metallocene complexes of the formula (IV), where the variables and indices are as defined in any of claims 1, 3 and 4.

9. An ansa-metallocene complex of the formula (I) as set forth in claim 1

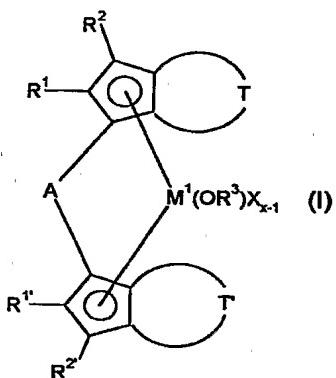


where the variables and indices are as defined in any of claims 1, 2 and 4.

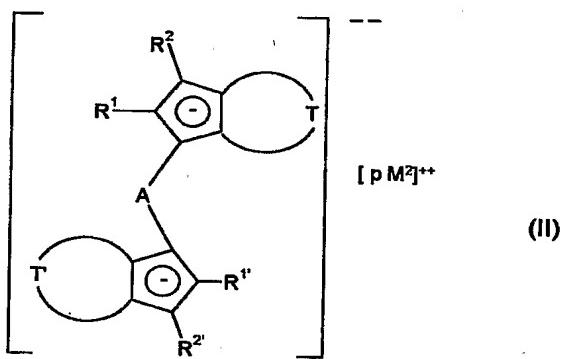
- 5 10. The use of an ansa-metallocene complex of the formula (I) prepared by a process as claimed in claim 1 as constituent of a catalyst system for the polymerization of olefins.

## **Abstract**

The present invention relates to a process for the meso-selective preparation of ansa-metallocene complexes of the formula (I).



which comprises reacting a ligand starting compound of the formula (II)



- 5 with a transition metal compound of the formula (III)



where

$R^1$ ,  $R^{1'}$  are identical or different and are each hydrogen or an organic radical having from 1 to 40 carbon atoms.

- 10  $R^2$ ,  $R^{2'}$  are identical or different and are each hydrogen or an organic radical having from 1 to 40 carbon atoms,  
 $R^3$  is a bulky organic radical which has at least 3 carbon atoms, is bound to the oxygen atom via a nonaromatic carbon or silicon atom and may be substituted by halogen atoms or further organic radicals having from 1 to 20 carbon atoms and may also contain heteroatoms selected from the group consisting of Si, N, P, O and S.

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T, T' are identical or different and are each a divalent organic group which has from 1 to 40 carbon atoms and together with the cyclopentadienyl ring forms at least one further saturated or unsaturated, substituted or unsubstituted ring system having a ring size of from 5 to 12 atoms, where T and T' may contain the heteroatoms Si, Ge, N, P, As, Sb, O,

5 S, Se or Te within the ring system fused to the cyclopentadienyl ring.

A is a bridge consisting of a divalent atom or a divalent group,

M<sup>1</sup> is an element of group 3, 4, 5 or 6 of the Periodic Table of the Elements or the lanthanides,

the radicals X are identical or different and are each an organic or inorganic radical which is able

10 to be replaced by a cyclopentadienyl anion,

x is a natural number from 1 to 4,

M<sup>2</sup> is an alkali metal, an alkaline earth metal or a magnesium monohalide fragment,

p is 1 in the case of doubly positively charged metal ions or 2 in the case of singly positively charged metal ions or metal ion fragments,

15 LB is an uncharged Lewis base ligand,

and

y is a natural number from 0 to 6,

and also the subsequent reaction of these complexes to form ansa-metallocenes of the formula (IV), the use of transition metal compounds of the formula (III) for preparing metallocenes and

20 also transition metal compounds of the formula (III), ansa-metallocene complexes of the formula (I) and the use of these as constituents of catalyst systems for the polymerization of olefins.